P.G. Semester-IV Examination, 2023 CHEMISTRY

Course ID: 41452 Course Code: CHEM-402E

Course Title: Organic Chemistry Special

Time: 2 Hours Full Marks: 40

The figures in the right-hand margin indicate marks.

Candidates are required to give their answers in their own words as far as practicable.

1. Answer any **five** of the following questions:

$$2 \times 5 = 10$$

a) Identify the missing reagents in the following reaction sequence:

b) Draw the structure of the missing reagent in the following reaction:

[Turn over]

- c) Write down the structure of a catecholate siderophores.
- d) 'The Norrish Type I process is mostly favoured by photolysis in the vapour phase and is less pronounced in the inert solvents'- Explain.
- e) Predict the product in the following reaction:

- f) Why lower oxidation state of chromium Cr(III) is used during oxidation?
- g) Write the adverse effect of Collin's reagent with example.
- 2. Answer any **four** of the following questions:

$$5 \times 4 = 20$$

a) i) Carry out the following transformation:

ii) Predict the products A to C in the following reaction sequence:

$$3+2=5$$

b) i) Write down the products formed in the following reaction and justify your answer:

Me
$$\frac{\text{IPCBH}_2}{\text{NaOH/H}_2O_2}$$
 A + E

- ii) Mention one limitation of IPC_2BH (Diisopinocampheylborane) in asymmetric hydroboration of an olefin. 4+1=5
- c) i) What do you mean by anticrowns? Give one example.
 - ii) Carry out the following transformation and show the mechanism involved:

$$(CH_2)_2COOMe$$
 $(CH_2)_2$
 $(CH_2)_2$
 $(CH_2)_6$

[3]

$$2+3=5$$

d) Predict the products formed in the following sequence of reactions with viable mechanism. 5

?
$$\frac{h\nu}{\text{solution}}$$
 $\frac{h\nu}{\ddot{H}}$?

- e) i) Write a short note on Paterno-Buchi reaction.
 - ii) Rationalize the reaction:

HO
$$NH_2$$
 $K_3[Fe(CN)_6]$ HO NH_2

$$3+2=5$$

f) How would you carry out the following transformations?

A.
$$C_6H_5$$
 C_6H_5

$$\mathsf{D}. \qquad \qquad \bigvee_{\hat{\mathsf{H}}}^{\mathsf{H}}$$

$$O_2N$$
 CHO O_2N CH_2OH

3. Answer any **on**e of the following questions:

$$10 \times 1 = 10$$

a) i) Predict the major products in the following reaction with suitable mechanism. Explain the selectivity, if any, involved.

- ii) Why molecular sieve is used in Sharpless asymmetric epoxidation (SAE) reaction?
- iii) Explain with suitable example the 'template effect' in the synthesis of macrocyclic compound. 5+1+4=10
- b) i) Identify the products formed in each of the following reactions:

ii) Write the appropriate reagent(s) for the following transformations:

A.
$$C_6H_5CH_2COOH$$

$$C_6H_5CH_1CH_2COOH$$

$$C_6H_{11}CH_1OH_2COOH$$